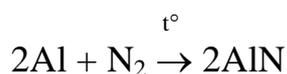
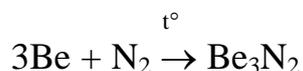
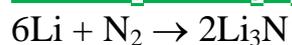
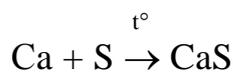
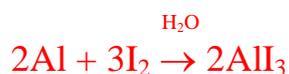
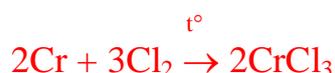
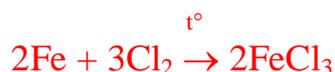
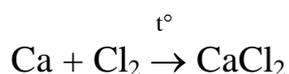
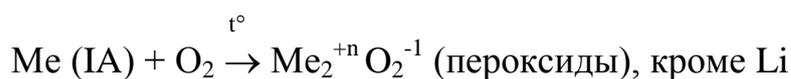
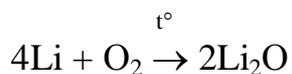
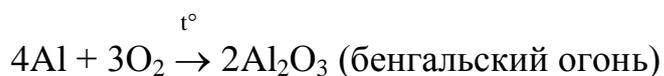


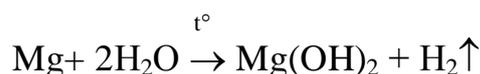
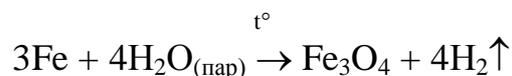
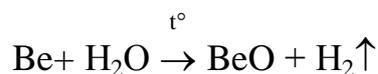
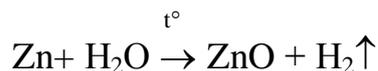
## Химические свойства металлов

### 1. Me + неMe → бинарное соединение

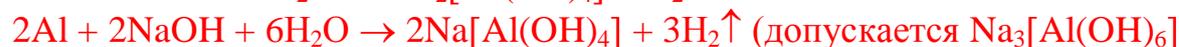




## другие Me с H<sub>2</sub>O



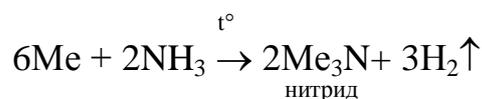
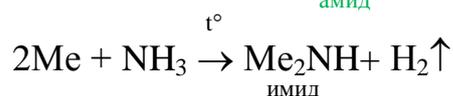
## 5. Me + щелочь + H<sub>2</sub>O → комплексное соединение + H<sub>2</sub>↑



## 6. Me + щелочь → соль + H<sub>2</sub>↑



## 7. 2Me + 2NH<sub>3</sub> → 2MeNH<sub>2</sub> + H<sub>2</sub>↑



## 8. Металлотермия

## 9. Me + орг.в-ва

(р.Вюрца, дегалогенирование дигалогеналканов, с алкинами по концевой тройной связи, со спиртами, фенолом, карбоновыми кислотами).