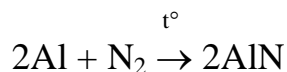
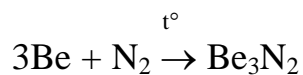
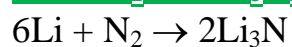
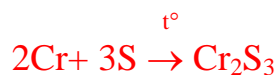
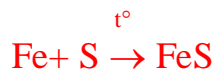
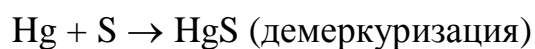
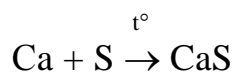
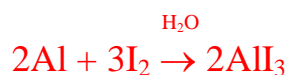
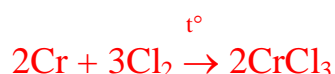
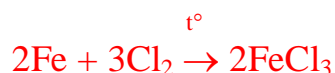
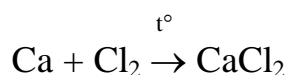
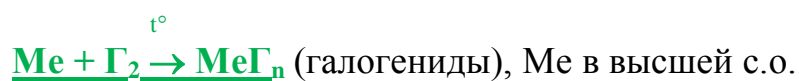
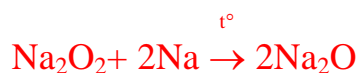
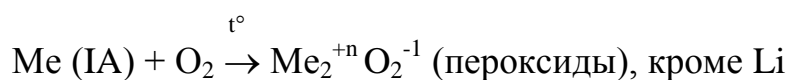
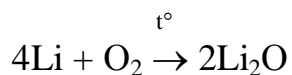
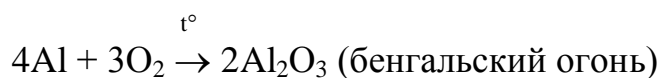
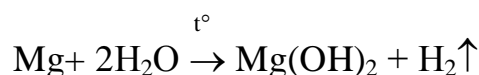
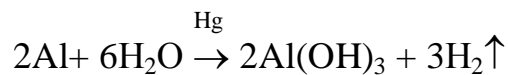
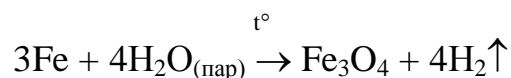
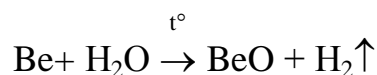
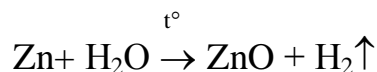


Химические свойства металлов

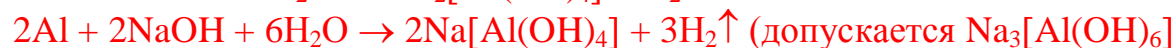
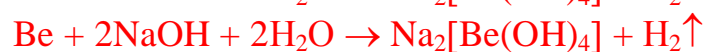
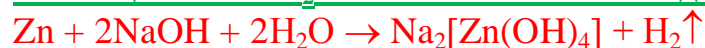
1. Me + неMe → бинарное соединение



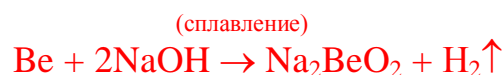
другие Me с H₂O



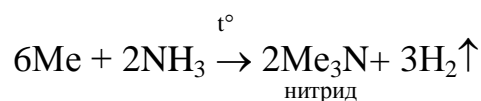
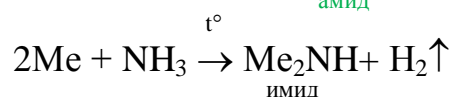
5. Me + щелочь + H₂O → комплексное соединение + H₂↑



6. Me + щелочь → соль + H₂↑



7. 2Me + 2NH₃ → 2MeNH₂ + H₂↑



8. Металлотермия

9. Me + орг.в-ва

(р.Вюрца, дегалогенирование дигалогеналканов, с алкинами по концевой тройной связи, со спиртами, фенолом, карбоновыми кислотами).